

MÖSSBAUER SPECTROSCOPY OF Fe-BASED NANOPOWDERS SYNTHESISED BY MICROWAVE PLASMA TORCH

B. David¹, N. Pizúrová¹, V. Kurdrle², O. Jašek², P. Synek²

¹Institute of Physics of Materials, ASCR, v.v.i., Brno, Czech Republic

²Faculty of Science, Masaryk University, Brno, Czech Republic

Microwave plasma torch at atmospheric pressure was used for the synthesis of Fe-based nanoparticles [1]. The discharge was ignited in argon which flowed through the central gas flow channel whereas the reactive mixture of H₂/O₂ gas and Fe(CO)₅ vapour was added by a concentric opening of the outer channel. The morphology and composition of the synthesised nanopowders were studied by TEM, XRD, Raman spectroscopy and Mössbauer spectroscopy.

Only γ -Fe₂O₃ ($d_{\text{XRD}} = 20$ nm) phase was identified by XRD in the **T89** sample. However the Fe₃O₄ characteristic peaks (669 cm⁻¹, 349 cm⁻¹) were also observed in the Raman spectrum.

γ -Fe₂O₃ ($d_{\text{XRD}} = 18$ nm, 41 wt.%), ε -Fe₂O₃ ($d_{\text{XRD}} = 27$ nm, 28 wt.%), and α -Fe₂O₃ ($d_{\text{XRD}} = 39$ nm, 31 wt.%) phases were identified in the XRD pattern of the **T104** sample. In its Raman spectrum the characteristic bands of α -Fe₂O₃ were observed besides the weak bands of γ -Fe₂O₃ and Fe₃O₄. The bands of ε -Fe₂O₃ were overlapped with the bands of α -Fe₂O₃, Fe₃O₄, and γ -Fe₂O₃.

In the **T107** sample α -Fe ($d_{\text{XRD}} = 48$ nm, 51 wt.%) and Fe₃O₄ ($d_{\text{XRD}} = 10$ nm, 49 wt.%) phases were found by XRD. The presence of Fe₃O₄ was clearly confirmed by Raman spectroscopy.

[1] L. Zajíčková, P. Eliáš, M., Jašek, O. et al. Plasma Phys. Control. Fusion 47 (2005) B655-B666

Presenting author: Bohumil David

Address: Žižkova 22, 61662 Brno, Czech Republic

FAX: +420 541 218 657

E-mail: david@ipm.cz